

FINAL EXAMINATION

ENG 310 – MASS TRANSFER FOR BIOLOGICAL

SYSTEMS

INSTRUCTION: ATTEMPT ALL QUESTIONS

TIME: THREE HOURS

Question 1. (20 Marks)

- a. Define Convective mass transfer and list the two different cases of convective mass transfer
- b. Discuss the significant parameters in connective mass transfer for the molecular diffusivities of the three transport processes (momentum, heat, and mass)

Question 2. (20 Marks)

- a. Define the following five terms
 - Mass
 - Velocity
 - Molecular weight

FINAL EXAMINATION

- Diffusion flux
- Drying

b. Discuss Fick's diffusional molar flux law and demonstrate Fick's rate equation in detail

Question 3. (20 Marks)

- a. Show the difference between diffusivity in liquid and diffusivity in solids giving examples for each case
- b. A vertical glass tube 3 mm in diameter is filled with liquid toluene to a depth of 20 mm from the top open end. After 275 hrs at 39.4°C and a total pressure of 760 mm Hg the level has dropped to 80 mm from the top. Calculate the value of diffusivity.

Question 4. (20 Marks)

- a. Oxygen is diffusing in a mixture of oxygen-nitrogen at 1 std atm, 25°C. The concentration of oxygen at planes 2 mm apart are 10 and 20 volume % respectively. Nitrogen is non-diffusing

FINAL EXAMINATION

- i. Derive the appropriate expression to calculate the flux oxygen. Define the units of each term clearly.
 - ii. Calculate the flux of oxygen. Diffusivity of oxygen in nitrogen = $1.89 \times 10^{-5} \text{ m}^2/\text{sec}$
- b. List four methods of operation of evaporators and draw a simplified well labeled diagram of one of the four choices you come up with

Question 5. (20 Marks)

- a. List and explain five types of dryers
- b. In a gas mixture of hydrogen and oxygen, steady equimolar counter diffusion is occurring at a total pressure of 100 kPa and temperature of 20°C . If the partial pressures of oxygen at two planes 0.01 m apart and perpendicular to the direction of diffusion are 15 kPa and kPa, respectively and the mass diffusion flux of oxygen in the mixture is $1.6 \times 10^{-5} \text{ kmol}/\text{m}^2 \cdot \text{sec}$, calculate the molecular diffusivity for the system.