

## Post Hartree-Fock Theory and Property Calculation

**Keywords:** Many body perturbation theory, Moller-Plesset, couple cluster, configuration interaction, Post Hartree Fock

### Post Hartree Fock Methods

In the last chapter, we have talked about a simple method to understand electron distribution in atoms and molecules known as Hartree-Fock (HF) theory. It is based on a particle independent model by taking all inter particle interactions into account in an average fashion, neglecting electron correlation. In this model, each electron is described by an orbital and the total wave function is given as a product of orbitals. The best set of orbitals is determined by the variational principle, i.e. the HF orbitals give the lowest energy within the restriction of the wave function being a single Slater determinant. In a sufficiently large basis set, the HF wave function is able to account for ~99% of the total energy. In order to calculate total energies of a molecular system with an accuracy of ~1 kcal/mol, it is necessary to use sophisticated methods for including electron correlation and large basis sets, which is computationally feasible only for small size systems. The difference in energy between the HF and the lowest possible energy applying a post-HF method in the given basis set is called the 'Electron Correlation' (EC) energy. Physically, it corresponds to the motion of the electrons being correlated resulting average inter particle distance to be more than that by the HF wave function.

Now we will move to examples where calculations are carried out applying post-HF electron correlated methods eg. many body perturbation theory (MBPT), couple cluster (CC) and configuration interaction (CI). At first, we take up an example to perform 'MBPT second order' calculations for geometry optimization of water molecule applying second order Moller-Plesset (MP2) correction.

### Moller-Plesset (MP2)

The following INPUT is prepared for such a calculation. The input is essentially same as in case of HF calculations adding extra keyword for MP2 method.

\*\*\*\*\*

```

$SYSTEM TIMLIM=600.0 MWORDS=10 MEMDDI=40 $END
$CONTRL SCFTYP=RHF ICHARG=0 MULT=1 RUNTYP=OPTIMIZE MPLEVEL=2 COORD=ZMT $END
$BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 NPFUNC=1 DIFFSP=.T. DIFFS=.T. $END
$DATA

```

Water-Geometry // MP2/6-31G(d,p)

CN 1

O  
 H 1 roh  
 H 1 roh 2 ahoh

It will perform only  
 HF calculations if  
 MPLEVEL=0 is supplied

roh=1.08  
 ahoh=109.5  
 \$END

\*\*\*\*\*

The final optimized geometrical parameters calculated applying MP2 method with the same geometrical input parameters and basis set as in case of HF theory are as follows:

\*\*\*\*\*

\*\*\*\*\* EQUILIBRIUM GEOMETRY LOCATED \*\*\*\*\*  
 COORDINATES OF ALL ATOMS ARE (ANGS)

ATOM	CHARGE	X	Y	Z
O	8.0	0.0000000000	-0.0672610890	0.0000000000
H	1.0	0.7661295166	0.5168495314	0.0000000000
H	1.0	-0.7661295166	0.5168495314	0.0000000000

THE CURRENT FULLY SUBSTITUTED Z-MATRIX IS

O  
 H 1 0.9634000  
 H 1 0.9634000 2 105.3549308

These are the  
 final geometrical  
 parameters

-----  
 MP2 ENERGY COMPONENTS  
 -----

WAVEFUNCTION NORMALIZATION = 1.0000000000

ONE ELECTRON ENERGY = -122.5342858206

TWO ELECTRON ENERGY = 37.1670578500

NUCLEAR REPULSION ENERGY = 9.1338520259

-----  
 TOTAL ENERGY = -76.2333759448

ELECTRON-ELECTRON POTENTIAL ENERGY = 37.1670578500

NUCLEUS-ELECTRON POTENTIAL ENERGY = -198.3822671061

NUCLEUS-NUCLEUS POTENTIAL ENERGY = 9.1338520259

-----  
 TOTAL POTENTIAL ENERGY = -152.0813572303

TOTAL KINETIC ENERGY = 75.8479812854

VIRIAL RATIO (V/T) = 2.0050811459

\*\*\*\*\*

To compare the calculated total energy of water molecule at HF and post-HF level of theory, energy components of neutral H<sub>2</sub>O molecule based on RHF optimization adopting the atomic basis functions are given below. Note that the energy values are given in a.u. (1 a.u. = 27.2114 eV = 627.5095 kcal/mol).

\*\*\*\*\*

-----  
**RHF ENERGY COMPONENTS**  
 -----

```

WAVEFUNCTION NORMALIZATION =          1.0000000000
      ONE ELECTRON ENERGY =        -123.3383142209
      TWO ELECTRON ENERGY =          37.9856765767
      NUCLEAR REPULSION ENERGY =         9.3290226296
      -----
      TOTAL ENERGY =          -76.0236150146

ELECTRON-ELECTRON POTENTIAL ENERGY =         37.9856765767
NUCLEUS-ELECTRON POTENTIAL ENERGY =        -199.2027162181
NUCLEUS-NUCLEUS POTENTIAL ENERGY =         9.3290226296
      -----
TOTAL POTENTIAL ENERGY =        -151.8880170118
TOTAL KINETIC ENERGY =          75.8644019972
VIRIAL RATIO (V/T) =           2.0020986525
  
```

\*\*\*\*\*

Thus, the calculated correlation energy of H<sub>2</sub>O molecule at MP2 level of theory with 6-31++G(d,p) basis set is (Energy at MP2 level - Energy at HF level) = (-76.2333759448 + 76.0236150146) a.u. = -0.2097609302 a.u. = ~ -5.708 eV ~ -131.63 kcal/mol (final values are truncated according to accuracy level).

Comparison of geometrical parameters of neutral H<sub>2</sub>O obtain at different level of theory with 6-31++G(d,p) basis set

Geometrical Parameter	Theoretical Method	
	RHF	MP2
rOH (Å)	0.9431015	0.9634000
<HOH (degree)	105.9642013	105.3549308

Note that for larger size systems, MP2 calculation will take significantly more time than HF procedure. Moreover, memory requirement will also be larger than HF calculation. So, one has to change the numbers

given to 'SYSTEM' group in the input header. The appropriate numbers may be generated by a test run for a set of geometrical data INPUT.

So far, all calculations in the previous examples are carried out for a neutral molecule with closed shell electronic structure. We will now consider water molecule with odd number of electrons i.e. by adding an extra electron to the molecule thus making an anion and by removing an electron from the molecule thus making a cation. For such a system, Unrestricted Hartee Fock (UHF) method has to be applied in place of Restricted Hartee Fock (RHF). The following two INPUT examples are prepared to UHF calculations on water anion and water cation systems.

\*\*\*\*\*

```

$SYSTEM TIMLIM=600.0 MWORDS=10 MEMDDI=40 $END
$CONTRL SCFTYP=UHF ICHARG=-1 MULT=2 RUNTYP=OPTIMIZE MLEVEL=0 COORD=ZMT $END
$BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 NPFUNC=1 DIFFSP=.T. DIFFS=.T. $END
$DATA
Water anion-Geometry // UHF/6-31G(d,p)
CN 1

O
H 1 roh
H 1 roh 2 ahoh

roh=1.08
ahoh=109.5
$END

```

\*\*\*\*\*

Note the changes made for keywords 'SCFTYP', 'ICHARG' and 'MULT' in 'CONTRL' group for this negatively charged water molecule keeping rest of the input same as the neutral one. As the system is an odd electron system, one has to choose UHF for SCF calculations. The negative charge of the system is given through the keyword, 'ICHARG=-1' and multiplicity of this doublet ( $s=1/2$ , multiplicity= $2s+1=2$ ) system is supplied by keyword, 'MULT=2'. The next INPUT is meant for water cation system. Again, it's an odd electron system, thus 'SCFTYP=UHF and MULT=2' are supplied. The keyword, 'ICHARG=1' is given to consider the system as positively charged one.

\*\*\*\*\*

```

$SYSTEM TIMLIM=600.0 MWORDS=10 MEMDDI=40 $END
$CONTRL SCFTYP=UHF ICHARG=1 MULT=2 RUNTYP=OPTIMIZE MLEVEL=0 COORD=ZMT $END
$BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 NPFUNC=1 DIFFSP=.T. DIFFS=.T. $END
$DATA
Water anion-Geometry // UHF/6-31G(d,p)
CN 1

```

```
O
H 1 roh
H 1 roh 2 ahoh
```

```
roh=1.08
ahoh=109.5
$END
```

```
*****
```

The final optimized geometrical parameters calculated applying UHF method with the same geometrical input parameters and basis set as in case of RHF method are as follows:

### For water anion system (UHF):

```
*****
```

```
***** EQUILIBRIUM GEOMETRY LOCATED *****
COORDINATES OF ALL ATOMS ARE (ANGS)
  ATOM   CHARGE      X           Y           Z
-----
O         8.0   0.0000000000  -0.2233870621  0.0000000000
H         1.0   0.8186920114   0.5841822048  0.0000000000
H         1.0  -0.8186920114   0.5841822048  0.0000000000
```

```
THE CURRENT FULLY SUBSTITUTED Z-MATRIX IS
```

```
O
H      1   1.1499673
H      1   1.1499673  2   90.7837318
```

These are the  
final geometrical  
parameters

```
-----
SPIN SZ   =   0.500
S-SQUARED =   0.814
-----
```

#### UHF ENERGY COMPONENTS

```
-----
WAVEFUNCTION NORMALIZATION =   1.0000000000
      ONE ELECTRON ENERGY =  -123.4975150138
      TWO ELECTRON ENERGY =   40.0633209235
      NUCLEAR REPULSION ENERGY =   7.6858601703
-----
TOTAL ENERGY =  -75.7483339201
```

```
ELECTRON-ELECTRON POTENTIAL ENERGY =   40.0633209235
NUCLEUS-ELECTRON POTENTIAL ENERGY =  -199.3951512057
```

```

NUCLEUS-NUCLEUS POTENTIAL ENERGY =      7.6858601703
-----
TOTAL POTENTIAL ENERGY =      -151.6459701119
TOTAL KINETIC ENERGY =      75.8976361918
VIRIAL RATIO (V/T) =      1.9980328469
    
```

\*\*\*\*\*

For water cation system (UHF):

\*\*\*\*\*

```

***** EQUILIBRIUM GEOMETRY LOCATED *****
COORDINATES OF ALL ATOMS ARE (ANGS)
ATOM   CHARGE      X              Y              Z
-----
O           8.0    0.0000000000   -0.2101249336   0.0000000000
H           1.0    0.8173304842    0.3347446215   0.0000000000
H           1.0   -0.8173304842    0.3347446215   0.0000000000
    
```

THE CURRENT FULLY SUBSTITUTED Z-MATRIX IS

```

O
H      1    0.9822993
H      1    0.9822993  2    112.6215572
    
```

These are the  
 final geometrical  
 parameters

```

-----
SPIN SZ   =    0.500
S-SQUARED =    0.757
-----
    
```

UHF ENERGY COMPONENTS

```

-----
WAVEFUNCTION NORMALIZATION =      1.0000000000

ONE ELECTRON ENERGY =      -117.6740287680
TWO ELECTRON ENERGY =      33.1026355691
NUCLEAR REPULSION ENERGY =      8.9431282670
-----
TOTAL ENERGY =      -75.6282649320

ELECTRON-ELECTRON POTENTIAL ENERGY =      33.1026355691
NUCLEUS-ELECTRON POTENTIAL ENERGY =      -193.0536141031
NUCLEUS-NUCLEUS POTENTIAL ENERGY =      8.9431282670
-----
TOTAL POTENTIAL ENERGY =      -151.0078502670
TOTAL KINETIC ENERGY =      75.3795853351
VIRIAL RATIO (V/T) =      2.0032990311
    
```

\*\*\*\*\*

Note the significant differences in geometrical parameters.

One has to also look into another important parameter,  $\langle s_z^2 \rangle$  with the calculated equilibrium geometrical parameters in the OUTPUT file. The values are **0.814** and **0.757** for negatively and positively charged water molecule at present. As you expect, the ideal  $\langle s_z^2 \rangle$  value for a **doublet** system should be **0.75**. The calculated large values are due to **spin contamination** that arises due to mixing of pure singlet and pure triplet states resulting artificial minima on UHF energy surface. This is a major drawback of UHF calculations. However, there are few procedures to bypass it and Restricted Open-shell Hartree Fock (ROHF) method is one such method. For a small system, one may not find any difference in CPU time to run a job under UHF and ROHF framework. However, for large system ROHF usually takes longer time compared to UHF method. Spin projection procedures are also available in some *ab initio* suit of packages to remove spin contamination from a wave function. With the same INPUT files for the negatively and positively charged water molecules with the keyword 'SCFTYP=ROHF', the following results are obtained.

### For negatively charged water molecule (ROHF):

\*\*\*\*\*

```

***** EQUILIBRIUM GEOMETRY LOCATED *****
COORDINATES OF ALL ATOMS ARE (ANGS)
  ATOM   CHARGE      X              Y              Z
-----
O         8.0    0.0000000000   -0.1179129403   0.0000000000
H         1.0    0.7905603808    0.5314451440   0.0000000000
H         1.0   -0.7905603808    0.5314451440   0.0000000000

```

THE CURRENT FULLY SUBSTITUTED Z-MATRIX IS

```

O
H      1   1.0230599
H      1   1.0230599  2   101.2013445

```

These are the  
final geometrical  
parameters

The calculated  $\langle s_z^2 \rangle$  value for this negatively charged **doublet** system is **0.750**, the ideal one.

### ENERGY COMPONENTS

```

-----
WAVEFUNCTION NORMALIZATION =          1.0000000000
      ONE ELECTRON ENERGY =         -125.2055682182
      TWO ELECTRON ENERGY =          40.7697568978
      NUCLEAR REPULSION ENERGY =          8.6106770234
-----
TOTAL ENERGY =          -75.8251342970
-----
ELECTRON-ELECTRON POTENTIAL ENERGY =          40.7697568978

```

```

NUCLEUS-ELECTRON POTENTIAL ENERGY = -201.3596224782
NUCLEUS-NUCLEUS POTENTIAL ENERGY = 8.6106770234
-----
TOTAL POTENTIAL ENERGY = -151.9791885570
TOTAL KINETIC ENERGY = 76.1540542600
VIRIAL RATIO (V/T) = 1.9956808608

```

\*\*\*\*\*

### For positively charged water molecule (ROHF):

\*\*\*\*\*

```

***** EQUILIBRIUM GEOMETRY LOCATED *****
COORDINATES OF ALL ATOMS ARE (ANGS)
ATOM   CHARGE      X           Y           Z
-----
O       8.0   0.0000000000  -0.2100155167  0.0000000000
H       1.0   0.8164757436   0.3346899131  0.0000000000
H       1.0  -0.8164757436   0.3346899131  0.0000000000

```

THE CURRENT FULLY SUBSTITUTED Z-MATRIX IS

```

O
H      1   0.9814971
H      1   0.9814971  2  112.5821473

```

These are the  
final geometrical  
parameters

Again, the calculated  $\langle s_z^{-2} \rangle$  value for this positively charged doublet system is **0.750**, the ideal one.

### ENERGY COMPONENTS

```

-----
WAVEFUNCTION NORMALIZATION = 1.0000000000
ONE ELECTRON ENERGY = -117.6894053944
TWO ELECTRON ENERGY = 33.1150833915
NUCLEAR REPULSION ENERGY = 8.9505117736
-----
TOTAL ENERGY = -75.6238102293
ELECTRON-ELECTRON POTENTIAL ENERGY = 33.1150833915
NUCLEUS-ELECTRON POTENTIAL ENERGY = -193.0636081665
NUCLEUS-NUCLEUS POTENTIAL ENERGY = 8.9505117736
-----
TOTAL POTENTIAL ENERGY = -150.9980130014
TOTAL KINETIC ENERGY = 75.3742027721
VIRIAL RATIO (V/T) = 2.0033115767

```

\*\*\*\*\*

Note that for charged systems, one should add diffuse functions to the chosen atomic basis functions for more accurate results. 'DIFFSP=.T. DIFFS=.T.' in INPUT (\$BASIS GBASIS=N31 NGAUSS=6 NDFUNC=1 NPFUNC=1 DIFFSP=.T. DIFFS=.T. \$END is equivalent to 6-31++G\*\*) ensures that diffuse functions are added to atomic basis functions. Now we will compare these results with the results based on [post-HF MP2](#) results under both UHF and ROHF frameworks and considering the same INPUT except keeping [MPLEVL=2](#) in 'CONTRL' group.

## Final MP2 results on negatively charged water molecule:

### UHF Framework

\*\*\*\*\*

```

***** EQUILIBRIUM GEOMETRY LOCATED *****
COORDINATES OF ALL ATOMS ARE (ANGS)
  ATOM   CHARGE      X              Y              Z
-----
O          8.0    0.0000000000  -0.0595114706   0.0000000000
H          1.0    0.7798623454   0.5129747222   0.0000000000
H          1.0   -0.7798623454   0.5129747222   0.0000000000

```

THE CURRENT FULLY SUBSTITUTED Z-MATRIX IS

```

O
H      1    0.9674325
H      1    0.9674325  2    107.4362421

```

```

SPIN SZ      =0.500
S-SQUARED   =0.750

```

#### ----- MP2 ENERGY COMPONENTS -----

```

WAVEFUNCTION NORMALIZATION =          1.0000000000
      ONE ELECTRON ENERGY =        -124.0815418643
      TWO ELECTRON ENERGY =          38.8259638483
      NUCLEAR REPULSION ENERGY =          9.0911379645
      -----
      TOTAL ENERGY =          -76.1644400515

ELECTRON-ELECTRON POTENTIAL ENERGY =          38.8259638483
NUCLEUS-ELECTRON POTENTIAL ENERGY =        -200.0001536701
NUCLEUS-NUCLEUS POTENTIAL ENERGY =          9.0911379645
      -----
      TOTAL POTENTIAL ENERGY =        -152.0830518573
      TOTAL KINETIC ENERGY =          75.9186118058
      VIRIAL RATIO (V/T) =          2.0032380498

```

\*\*\*\*\*

ROHF Framework

\*\*\*\*\*

```

***** EQUILIBRIUM GEOMETRY LOCATED *****
COORDINATES OF ALL ATOMS ARE (ANGS)
  ATOM   CHARGE      X              Y              Z
-----
O         8.0    0.0000000001  -0.0866386393  0.0000000000
H         1.0    0.7599569624   0.5265383065  0.0000000000
H         1.0   -0.7599569625   0.5265383066  0.0000000000

```

THE CURRENT FULLY SUBSTITUTED Z-MATRIX IS

```

O
H      1    0.9764838
H      1    0.9764838  2    102.2028330

```

```

-----
SPIN SZ      =      0.500
S-SQUARED   =      0.750
-----

```

```

-----
MP2 ENERGY COMPONENTS
-----

```

```

WAVEFUNCTION NORMALIZATION =      1.0000000000
      ONE ELECTRON ENERGY =     -124.2678996698
      TWO ELECTRON ENERGY =      39.0499185082
      NUCLEAR REPULSION ENERGY =      9.0189017320
      TOTAL ENERGY =     -76.1990794297
ELECTRON-ELECTRON POTENTIAL ENERGY =      39.0499185082
NUCLEUS-ELECTRON POTENTIAL ENERGY =    -200.1567551801
NUCLEUS-NUCLEUS POTENTIAL ENERGY =      9.0189017320
      TOTAL POTENTIAL ENERGY =    -152.0879349400
      TOTAL KINETIC ENERGY =      75.8888555103
      VIRIAL RATIO (V/T) =      2.0040878719

```

\*\*\*\*\*

**Final MP2 results on positively charged water molecule:**UHF Framework

\*\*\*\*\*

```

***** EQUILIBRIUM GEOMETRY LOCATED *****
COORDINATES OF ALL ATOMS ARE (ANGS)
  ATOM   CHARGE      X           Y           Z
-----
O         8.0    0.0000000000  -0.0589999594  0.0000000000
H         1.0    0.8212678720   0.5127189666  0.0000000000
H         1.0   -0.8212678720   0.5127189666  0.0000000000

```

THE CURRENT FULLY SUBSTITUTED Z-MATRIX IS

```

O
H      1  1.0006715
H      1  1.0006715  2  110.3132177

```

```

SPIN SZ    =0.500
S-SQUARED =0.757

```

-----  
MP2 ENERGY COMPONENTS  
-----

```

WAVEFUNCTION NORMALIZATION =      1.0000000000
      ONE ELECTRON ENERGY =     -117.1910616872
      TWO ELECTRON ENERGY =      32.6322015518
      NUCLEAR REPULSION ENERGY =      8.7833252537
      TOTAL ENERGY =     -75.7755348817
ELECTRON-ELECTRON POTENTIAL ENERGY =      32.6322015518
NUCLEUS-ELECTRON POTENTIAL ENERGY =    -192.5669646924
NUCLEUS-NUCLEUS POTENTIAL ENERGY =      8.7833252537
      TOTAL POTENTIAL ENERGY =    -151.1514378869
      TOTAL KINETIC ENERGY =      75.3759030052
      VIRIAL RATIO (V/T) =      2.0053018519

```

\*\*\*\*\*

ROHF Framework

\*\*\*\*\*

```

***** EQUILIBRIUM GEOMETRY LOCATED *****
COORDINATES OF ALL ATOMS ARE (ANGS)
  ATOM   CHARGE      X           Y           Z
-----
O         8.0    0.0000000000  -0.0589906126  0.0000000000
H         1.0    0.8210594088   0.5127142932  0.0000000000
H         1.0   -0.8210594088   0.5127142932  0.0000000000

```

THE CURRENT FULLY SUBSTITUTED Z-MATRIX IS

```

O
H      1   1.0004924
H      1   1.0004924  2  110.3008942

```

```

-----
SPIN SZ   =   0.500
S-SQUARED =   0.750
-----

```

```

-----
MP2 ENERGY COMPONENTS
-----

```

```

WAVEFUNCTION NORMALIZATION =      1.0000000000
      ONE ELECTRON ENERGY =     -117.1942100576
      TWO ELECTRON ENERGY =      32.6344646038
      NUCLEAR REPULSION ENERGY =      8.7849216446
-----
TOTAL ENERGY =     -75.7748238092
-----
ELECTRON-ELECTRON POTENTIAL ENERGY =      32.6344646038
NUCLEUS-ELECTRON POTENTIAL ENERGY =     -192.5698192565
NUCLEUS-NUCLEUS POTENTIAL ENERGY =      8.7849216446
-----
TOTAL POTENTIAL ENERGY =     -151.1504330081
TOTAL KINETIC ENERGY =      75.3756091989
VIRIAL RATIO (V/T) =      2.0052963368

```

\*\*\*\*\*

As the total energy of neutral, negatively and positively charged water molecule have been calculated electron affinity (EA) and ionization potential (IP) of H<sub>2</sub>O molecule can be calculated applying following relations.

Electron affinity (EA) = Total energy (neutral) – Total energy (negatively charged)

Ionization potential (IP) = Total energy (positively charged) – Total energy (neutral)

The calculated EA and IP values of H<sub>2</sub>O molecule under gas phase isolated condition at different levels of theory are tabulated below for comparison.

Calculated Parameter	Theoretical Method			
	HF		Post HF (MP2)	
	RHF/UHF	RHF/ROHF	RHF/UHF	RHF/ROHF
EA (eV)	7.49	5.40	1.87	0.93
IP (eV)	10.76	10.88	12.45	12.48

Experimental value: EA = 1.3 eV and IP = 12.61 eV

One can see from the above Table that post-HF MP2 method is capable of predicting electron affinity (EA) and ionization potential (IP) values quite close to experimental ones. It may improve even further adopting a large atomic basis function. MP2 method typically accounts for 80–90% of the correlation energy and the third order energy correction (MP3) typically accounts for 90–95% of the correlation energy. The full fourth order energy correction (MP4) accounts for 95–98% of the correlation energy. However, computational cost increases significantly from MP2 to MP4 method. It is prescribed to perform geometry optimization at MP2 level of theory and energy of the system may be improved further by MP4 procedure. We will also discuss other post-HF methods which is capable of producing more accurate energy parameters.

### Couple Cluster (CC)

Many Body Perturbation Theories (MBPT) add different types of corrections to the reference wave function to a given order (second, third or fourth as in MP2, MP3 or MP4). The idea in Coupled Cluster (CC) methods is to include all corrections of a given type to infinite order. Coupled cluster is closely connected with Moller-Plesset perturbation theory and the infinite Taylor expansion of the exponential operator ensures that the contributions from a given excitation level are included to infinite order. Perturbation theory indicates that doubles are the most important, since they are the only contributors to MP2 and MP3. At fourth order, there are contributions from singles, doubles, triples and quadruples. In the '\$CONTRL' group if 'CCTYP=CCD' is used GAMESS performs a CC calculation using the coupled-cluster method with doubles which is the lowest level of approximation. Whereas for 'CCTYP=CCSD' it performs coupled-cluster calculations with singles and doubles. When, the triples contribution are evaluated by perturbation theory and added to the CCSD results it is known as CCSD(T) procedure. This is capable of predicting very accurate results accounting 99.99% correlation energy and applicable only for medium size molecules due to computational cost.

The calculated couple cluster energies with different levels of corrections along with MP2 method are listed below for comparison. The calculations are carried out with the same set of geometrical parameters of neutral H<sub>2</sub>O molecule.

#### SUMMARY OF RESULTS

REFERENCE ENERGY:	-76.0304191949	
MBPT(2) ENERGY:	-76.2333759470	CORR. E= -0.2029567522
CCSD ENERGY:	-76.2412017458	CORR. E= -0.2107825509
CCSD(T) ENERGY:	-76.2446812103	CORR. E= -0.2142620154

'REFERENCE ENERGY' refers to the HF energy of H<sub>2</sub>O molecule with the same set of geometrical parameters and 'CORR. E' refers to the calculated 'electron correlation energy' at different level of theory. This method is used mainly for single point energy calculation as it is computationally expensive.

#### Configuration Interaction (CI)

In this procedure, the trial wave function is written as a linear combination of determinants. Determinants that are singly, doubly, triply, etc. excited relative to the HF configuration and the energy is minimized under the constraint that the total CI wave function is normalized. The excited Slater determinants are generated by removing electrons from occupied orbitals, and placing them in virtual orbitals. The number of excited Slater Determinants increases factorially with the number of electrons and basis functions. This is the best possible wave function within the limitations of the basis set, i.e. it recovers 100% of the electron correlation in the given basis. However, even for a system like H<sub>2</sub>O molecule (18 electrons) with a 6-31G\* basis set it is impossible to solve. In order to develop a computationally tractable model, the number of excited determinants in the CI expansion must be reduced. Truncating the excitation level to one (known as CI singles or CIS) does not give any improvement over the HF result. The lowest CI level that gives an improvement over the HF result is to include only doubly excited states, yielding the CI with doubles. Compared with the number of doubly excited determinants, there are relatively few singly excited determinants and including these gives the CISD (CI singles and doubles) method. This is a method commonly used for small to medium sized molecules.

#### Summary

In this chapter, discussions on post-HF method, MP2 are made and calculated results are compared with HF method both for closed and open shell systems. Calculated results on EA and IP are also compared with available experimental data. Discussions on CC and CI method are also made with reference to accuracy of results and computational cost.