

### Computing spin-dependent properties

If the molecule under consideration is not in a spin singlet electronic state ( $s \neq 0$ ), it carries a net non-zero electronic spin angular momentum. If some of the nuclei in the molecule carries non zero nuclear spin, the electronic spin can couple with the non-zero nuclear spin. The energetic of this coupling depends on whether electronic spin, projected along the z-axis is aligned with the nuclear spin projected on the same axis or is in opposition. The difference in the energies corresponding to the two possibilities is probed in electron spin resonance spectroscopy. The measurement leads the extraction of isotopic hyperfine splitting or equivalently as hyperfine coupling constant ( $a_x$ ) of the nucleus x. The theoretical calculation of  $a_x$  is fairly straight forward and leads to the expression.

$$a_x = \frac{4}{3} \pi < s_z >^{-1} g \beta g_x \beta_x \rho(x) \dots \dots \dots (23.1)$$

Where  $g, \beta$  are the electronic 'g' factor and the Bohr magneton respectively.  $g_x, \beta_x$  are the corresponding quantities for the nuclei designated as x.  $< s_z >$  is the average value of the z-component of the electronic spin angular momentum and  $\rho(x)$  is the Fermi contact integral. If the wavefunction  $\Psi$  is represented as a single Slater determinant, the  $\rho(x)$  can be written as,

$$\rho(x) = \sum_{\mu\nu} P_{\mu\nu}^{spin} \{ \chi_\mu(r_x) \chi_\nu(r_x) \} \dots \dots \dots (23.2)$$

In equation (2)  $\{ \chi_\mu(r_x) \chi_\nu(r_x) \}$  is the overlap of the basis functions,  $\chi_\mu$  and  $\chi_\nu$  at the nucleus x.  $P_{\mu\nu}^{spin}$  is the  $\mu\nu$ <sup>th</sup> element of the spin density matrix constructed in the AO basis.

If the MOs carrying electrons of  $+\frac{1}{2}$  spin ( $\alpha$ -spin) and  $-\frac{1}{2}$  spin ( $\beta$ -spin) are expressed

separately as linear combination of the atomic orbital basis set we have,

$$(\phi_1^\alpha, \phi_2^\alpha, \dots, \phi_{N_\alpha}^\alpha) \equiv \phi(\alpha) = (\chi_1, \chi_2, \dots, \chi_N) C_\alpha \quad (23.3)$$

$$(\phi_1^\beta, \phi_2^\beta, \dots, \phi_{N_\beta}^\beta) \equiv \phi(\beta) = (\chi_1, \chi_2, \dots, \chi_N) C_\beta \quad (23.4)$$

Where  $C_\alpha$  is the  $N \times N_\alpha$  dimensional co-efficient matrix of expansion co-efficient of the  $\alpha$ -spin molecular orbitals, while  $C_\beta$  is  $N \times N_\beta$  dimensional co-efficient matrix of the  $\beta$ -spin molecular orbitals ( $C_\alpha^\dagger S C_\alpha = 1, C_\beta^\dagger S C_\beta = 1, S$  being the  $N \times N$  overlap electrons in the AO basis given by,

$$P^\alpha = C_\alpha C_\alpha^\dagger \quad (23.5)$$

$$P^\beta = C_\beta C_\beta^\dagger \quad (23.6)$$

The spin-density matrix  $P^{\text{spin}}$  is the difference between  $P^\alpha$  and  $P^\beta$

$$P^{\text{spin}} = (P^\alpha - P^\beta) \quad (23.7)$$

If  $S=0$ , and each spatial orbital is occupied by two electrons of opposite spin,  $P^\alpha = P^\beta$  and  $P^{\text{spin}}=0$  leading to  $a_x=0$ . If,  $S \neq 0$  ( $=\frac{1}{2}, \frac{3}{2}$ , etc.),  $P^\alpha \neq P^\beta$ ,  $P^{\text{spin}} \neq 0$ ,  $a_x \neq 0$ .

In the restricted Hartree-Fock description the spatial orbitals used to describe electrons of  $\alpha$  and  $\beta$ -spin are identical. So, the contribution to  $P^{\text{spin}}$  comes only from the orbitals occupied singly. It may so happen that the singly occupied MO has a nodal plane passing through the atom 'x'. In that case the  $a_x$  value would be predicted to be zero in the RHF description. This would be wrong as apart from the delocalization contribution, spin density is also created by spin-polarization which is not accounted for in RHF theories.

The UHF method assigns different spatial orbitals to  $\alpha$  and  $\beta$ -spin electrons. Hence  $P^{\text{spin}}$  naturally contains contribution from spin polarization as well as delocalization effects. The problem in UHF is that  $\Psi_{\text{UHF}}$  is not an eigenfunction of the  $S^2$  operator so that the doublet will have contaminations from spin quartet and states with still higher spin multiplicities. (Triplet will have contaminations from quintet, etc.)

One must therefore project out higher spin components and the projected UHF spin density matrix should be used in the computation of  $P^{\text{spin}}$ . Projection after variational optimization is usually done, though it is better to perform it before optimization.

Note:

1. The expression for the Fermi contact integrals show that  $P_{\mu\nu}^{\text{spin}}$  is multiplied with the overlap of the basis functions  $\chi_\mu(x)$ , and  $\chi_\nu(x)$  at the nuclear position  $x$ . The maximum amplitude of the basis function at nuclear position ( $x$ ) comes from the s-type atomic orbitals in the basis. Unfortunately the core orbitals are not usually endowed with the same degree of flexibility as the valence orbitals in the basis and this could affect the computed coupling constant ( $a_x$ ).
2. The nuclear cusp behavior is better represented when Slater type of atomic orbitals are used. The GTOs are relatively unsatisfactory in this respect.

3. The derivation of equation (1) follows easily from the application of the Hellmann-Feynman theorem (HFT) to the molecular electronic Hamiltonian containing an addition perturbation arising from the interaction between the electronic spin and nuclear spin magnetic moments.
4. At the post-Hartree-Fock level, both MP2 and DFT calculations are recommended for the calculation of  $\alpha_X$  values.